Amino Acid Catabolism and Generation of Volatiles by Lactic Acid Bacteria

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ABSTRACT

Twelve isolates of lactic acid bacteria, belonging to the Lactobacillus, Lactococcus, Leuconostoc, and Enterococcus genera, were previously isolated from 180-d-old Serra da Estrela cheese, a traditional Portuguese cheese manufactured from raw milk and coagulated with a plant rennet. These isolates were subsequently tested for their ability to catabolize free amino acids, when incubated independently with each amino acid in free form or with a mixture thereof. Attempts were made in both situations to correlate the rates of free amino acid uptake with the numbers of viable cells. When incubated individually, leucine, valine, glycine, aspartic acid, serine, threonine, lysine, glutamic acid, and alanine were degraded by all strains considered; arginine tended to build up, probably because of transamination of other amino acids. When incubated together, the degradation of free amino acids by each strain was dependent on pH (with an optimum pH around 6.0). The volatiles detected in ripened Serra da Estrela cheese originated mainly from leucine, phenylalanine, alanine, and valine, whereas in vitro they originated mainly from valine, phenylalanine, serine, leucine, alanine, and threonine. The wild strains tested offer a great potential for flavor generation, which might justify their inclusion in a tentative starter/nonstarter culture for that and similar cheeses.

(Key words: flavor, catabolism, starter cultures, cheese)

INTRODUCTION

Most flavor notes that specifically characterize each type of cheese develop during ripening, due to the presence of a number of actively metabolizing microorganisms (with starter and nonstarter roles), as well as enzymes secreted by these microorganisms (or released therefrom after lysis) coupled with enzymes indigenous in milk (or added thereto as part of the rennet) (Fox and Law, 1991). The main sapid compounds in cheese are produced through the primary reactions of glycolysis, lipolysis, and proteolysis. Such compounds, especially free amino acids (FAA), do often undergo further transformation via several secondary reactions which are still poorly known (Fox and Wallace, 1997), to eventually yield aroma compounds. Consequently, the actual concentration of the former compounds in cheese has been used as a basis for the calculation of ripening indices, as it usually correlates well with flavor intensity.

The dominant FAA in Cheddar cheese are Leu, Glu, and Lys (Wood et al., 1985), although variations in their relative amounts have been reported by other authors (Broome et al., 1990). In European traditional cheeses produced from raw ewe’s and/or goat’s milk (e.g., Manchego cheese), the amounts of almost every type of FAA increase with ripening time (Ordoñez and Burgos, 1980). In Idiazabal cheese, increases in the total FAA concentration from 200 mg per 100 g of DM (mg/100 gDM) in 1-d-old to 1300 mg/100 gDM in 12-mo-old cheeses have been reported (Barcina et al., 1995); the dominant FAA found were Glu, Val, Leu, and Phe. Studies pertaining to Portuguese traditional cheeses, e.g., Picante da Beira Baixa, have also pointed at increases of total FAA as ripening time elapses, say from approximately 30 to 54 mg/100 gDM at 0 d to 1330 to 1940 mg/100 gDM by 180 d (Freitas et al., 1998); the dominant FAA were, in this case, Val, Leu, and Phe.

Lactic acid bacteria (LAB) are nutritionally fastidious microorganisms, which are, nevertheless, able to
hydrolyze milk peptides down to FAA. Amino acid catabolism produces, in turn, a number of compounds, including ammonia, amines, aldehydes, phenols, indole and alcohols, which contribute as a whole to cheese flavor (Urbach, 1995); however, the roles played by each species (or even genus) of LAB in terms of those biochemical routes are not yet fully understood. There are usually three recognizable steps in this complex process: the first one pertains to such reactions as decarboxylation, deamination, transamination, desulfuration, and hydrolysis of side-chains; the second one involves conversion of the resulting compounds (mainly amines and α-ketoacids), as well as some FAA themselves, to aldehydes affected by deaminases; and the third stage corresponds to reduction of aldehydes to alcohols, or their oxidation to carboxylic acids. Sulfur-containing FAA may undergo specific chemical reactions, which are responsible mainly for the generation of methanethiol and a few other sulfur derivatives.

Volatiles evolving from the metabolism of FAA do apparently contribute to the final organoleptic characteristics of ripened cheeses. Extensive reviews (Urbach, 1995; Fox and Wallace, 1997) are available pertaining to the profile of volatiles in many cheese varieties; some studies have even focused on microbial-mediated volatile components, derived from individual FAA, in homogeneous model systems. Improved mass spectrometric methods of detection after resolution by gas chromatography can be applied to the analysis of volatiles in cheese extracts; these methods have contributed considerably to the accumulated body of knowledge encompassing flavor compounds (Bosset and Gauch, 1993; Esteban et al., 1996). Nevertheless, the relationship between the numbers of viable microorganisms that may produce such volatiles and the concentration of said volatiles is not trivial. Efforts to unfold such a relation require, as a primary approach, that the metabolic potential of each microbial species be investigated independently in a model system.

In addition to the aforementioned limitation, studies on Portuguese traditional cheeses, known for their unique bouquet, are even scarcer. A plethora of new studies have even focused on microbial-mediated volatile components, derived from individual FAA, in homogeneous model systems. Improved mass spectrometric methods of detection after resolution by gas chromatography can be applied to the analysis of volatiles in cheese extracts; these methods have contributed considerably to the accumulated body of knowledge encompassing flavor compounds (Bosset and Gauch, 1993; Esteban et al., 1996). Nevertheless, the relationship between the numbers of viable microorganisms that may produce such volatiles and the concentration of said volatiles is not trivial. Efforts to unfold such a relation require, as a primary approach, that the metabolic potential of each microbial species be investigated independently in a model system.

Therefore, our study intended to shed light into this particular area by investigating the catalytic potential of LAB that had been previously isolated from Serra da Estrela cheese. For such goal, a twofold approach was followed, based on either individual FAA or all FAA in a cocktail, at various pH values (4.5, 6.0, and 7.5). Curdled milk was also tested with four of the isolates in order to approach cheese-like conditions.

**MATERIALS AND METHODS**

**Microorganisms Employed**

Twelve colonies of LAB, previously isolated from 180-d-old Serra da Estrela cheese, were used in this study. Three of them, namely ESB 321, ESB 322, and ESB 323, were isolated from Rogosa agar (Merck, Darmstadt, Germany) and identified by SDS-PAGE as *Lactobacillus paracasei* ssp. paracasei, *Lactobacillus brevis* and *Lactobacillus rhamnosus/plan tarum*, respectively. Isolates ESB 326, ESB 327, and ESB 330 were isolated from Mayeux, Sandine, and Elliker (MSE) agar (Biokar, Beauvais, France) and identified by SDS-PAGE as *Leuconostoc mesenteroides ssp. dextranicum* (first two) and *Leuconostoc lactis* (last one). From M17 agar, the following species (identified by SDS-PAGE) were also isolated: *L. lactis ssp. lactis* (ESB 331), *L. brevis* (ESB 332), and *L. lactis* (ESB 333). Finally, from KF-Streptococcus agar (Merck), the isolates were identified by SDS-PAGE as *Enterococcus faecium* (ESB 337 and 338) and *E. faecalis* ssp. *liquefaciens* (ESB 340). The chemotaxonomical method employed, based on SDS-PAGE of disrupted cultures followed by dendogram grouping, was described to full length by Kersters (1985) and Jackman (1985).

**Buffer Experiments**

Selected isolates, kept on MRS (*Lactobacillus* and *Leuconostoc*) or M17 (*Lactococcus* and *Enterococcus*) agar slants, were subcultured in MRS or M17 (Merck) broth as appropriate (cultures were thus maintained in general purpose, rather than selective media, as regular practice in applied microbiology). They were then inoculated at 2% in 100 mL of fresh MRS broth and allowed to reach the stationary phase (∼16 h); after this, they were centrifuged at 12,000 × g for 20 min using a Sorvall Centrifuge RC5 (Wilmington, DL) and washed twice with phosphate buffer (pH 6.0). The pellet was resuspended in 56 mL of phosphate buffer (50 mM, pH 6.0), added to 44 mL of buffer containing 0.02 % (wt/vol) of the desired FAA, and incubated at 30°C for up to 7 d. The phosphate buffer added to the washed cells...
contained either a single FAA or a mixture of 16 FAA (25 mM), without or with added α-ketoglutarate (10 mM) (Sigma Chemical, St. Louis, MO).

Curdled Milk Experiments

Sterilized sheep’s milk curdled with C. cardunculus (aqueous crude extract, containing 0.3 g/L of dry flowers and 12 g/L of NaCl) was inoculated (2% of washed cell suspension, prepared as described above) with each of the four isolates individually (ESB 321, ESB 327, ESB 331, and ESB 338), to follow FAA degradation in the absence or presence of added 0.025 mol/L of α-ketoglutarate. Samples were incubated at 30°C (Lactobacillus and Leuconostoc) or 37°C (Lactococcus and Enterococcus). The concentration of the FAA remaining and the number of viable cells were monitored at the following time intervals: 0, 15, 30, and 45 d. Sterilized sheep’s milk, curdled with C. cardunculus and not inoculated at all, was used as control.

Microbiological Analyses

Samples (1 ml) from the buffer experiments and the curdled milk experiments were taken at selected time intervals, say 0, 3, 5, and 7 d, and 0, 15, 30, and 45 d, respectively, and decimally diluted in sterile 0.1% (wt/vol) peptone water (Sigma). They were then plated in duplicate on MRS agar, to assay for: Lactobacillus, as standard practice; and for Leuconostoc, following the successful results by Macedo et al. (1995) (results not shown have been generated using M17—a medium recommended for lactic streptococci—but no significant differences were found); and on M17 agar, to assay for Lactococcus and Enterococcus. In both cases, the spread-plate technique was used. The plates were incubated anaerobically (Lactobacillus and Lactococcus) and aerobically (Leuconostoc and Enterococcus), at 30°C for 3 to 4 d (Tavaaria and Malcata, 1998); after this time, the colonies were duly counted.

Analysis of Free Amino Acids

In experiments where single FAA were used in each incubation, samples (1.0 ml), previously cleared of cells and cell debris by centrifugation for 15 min at 3,000 × g, were added to 2 ml of Cd-ninhydrin reagent; this reagent was prepared by first dissolving 0.8 g of ninhydrin (Sigma) in a mixture of 80 ml of absolute ethanol (Romil, Loughborough, UK) and 10 ml of acetic acid (Pronalab, Lisbon, Portugal), followed by addition of 1 g of CdCl₂ (Sigma) dissolved in 1 ml of distilled water. The mixture was heated at 84°C for 5 min and then cooled to room temperature; absorbance was read at 507 nm, except for proline for which absorbance was read at 440 nm (Folkertsma and Fox, 1992). A calibration curve was constructed for each FAA. To determine FAA in samples containing a mixture thereof, either in buffer or in curdled milk, 0.2 ml of the sample was taken, centrifuged for 15 min at 3,000 × g, and mixed with 200 μl of 9-fluorenyl-ethyl-chloroformate, FLEC (Sigma-Aldrich, Steinheim, Germany) for 5 min. The excess reagent was eliminated by addition of 200 μmol of 2 M amantadine (Sigma-Aldrich) in acetone. The mixture was then centrifuged for 15 min at 3,000 × g to remove the precipitate; 5 μl of the supernatant was finally injected in the HPLC system (LaChrom, Merck, Germany) using an automatic injection system. Separation was accomplished at 30°C via a ChromSep Microsphere C18 column (50 mm × 4.6 mm × 4 μm), and detection was carried out using UV absorption at 254 nm. The flow rate was 2.0 ml/min; the mobile phases were: acetonitrile (A), methanol (B) and buffer (3 ml of acetic acid and 1 ml of NH₄OH/L of ultrapure water) (C). The following gradient sequence was used: A: 10% for all times; B: 15, 30, 45, 60, and 75% at 0, 2, 5, 9, and 14 min, respectively; C: 75, 60, 45, 30, and 15 min at 0, 2, 5, 9, and 14 min, respectively. [AQ: Verify wording and insertion of (Table 1).]

Calibration was previously done by injecting standard solutions of chromatographic grade FAA (Sigma), at several concentrations.

Analysis of Volatiles

Volatiles resulting from in vitro FAA catabolism were recovered using solid-phase microextraction, a technique pioneered by Chin et al. (1996). Headspace sampling was carried out using N₂ to carry the volatiles released from the sample onto a polyacrylate fiber, which they were eventually adsorbed to. Soon after this, adsorbed volatiles were desorbed via a thermal shock up to the temperature of the injector in the gas chromatograph (Thermo Quest, Trace GC 2000 Series; Finnigan, San Jose, CA) in splitless mode. Separation was achieved in a DB-Wax column (60 m × 0.32 mm × 0.25 μm) from J&W Scientific (Folsom, CA). Gas chromatography-mass spectrometry was carried out with the same column using a GCQ MS detector from Thermo Quest (Finnigan). The oven temperature was maintained at the initial value of 35°C for 2 min, then increased to 175°C (at 4°C/min), and finally increased to 225°C (at 10°C/min). System control, as well as data acquisition and analysis, were all accomplished with data processing and analysis software from Finnigan.

Statistical Analyses

ANOVA was conducted, at the 5% level of significance, for the FAA concentration (curdled milk experi-
Table 1. Percent degradation of FAA, after individual incubation at 30°C for 7 d with selected strains of LAB. Negative values indicate increases in concentration.

<table>
<thead>
<tr>
<th>Strain</th>
<th>Lys</th>
<th>Thr</th>
<th>Trp</th>
<th>Asp</th>
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<th>Ala</th>
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<th>Arg</th>
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nd = not determined.

RESULTS AND DISCUSSION

Metabolism of Free Amino Acids Independently

The rate of degradation of every single FAA was monitored via incubation of individual FAA in phosphate buffer with selected wild strains, previously isolated from ripened Serra da Estrela cheese. It was somewhat expected that these isolates would preferably degrade those FAA most abundant in that cheese (Leu, Val, Glu, and Lys) at the later stages of maturation. However, this was not the case of our experimental observations. In fact, some of the less prevalent FAA (e.g., Ser) were totally degraded by some isolates. From the major FAA prevalent in Serra da Estrela cheese by 180 d of ripening, and which were individually subject to action by each of the selected isolates (Table 1), it can be concluded that Lys and Glu were completely degraded by all. Valine was degraded up to 59.4%, whereas Leu was degraded up to 66.8%. In what pertains to the other FAA, Asp was substantially degraded by all strains (from 31 to up to 100%), as well as Ser (from 27 to up to 100%), Thr (from 33.5 to up to 100%), and Ala (from 93.6 to up to 100%). The concentration of Arg tended to build up in 10 out of 12 isolates. A clear trend could not be found for these strains toward Trp, Ile, Met, Cys, and His. When studying FAA catabolism by four strains of Propionibacterium, Brendehaug and Langsrud (1985) also reported large differences among strains in their ability to degrade FAA.

Metabolism of Free Amino Acids in Cocktail

Figure 1 represents amino acid degradation by four of the isolates considered in this study, at various pH values (4.5, 6.0, and 7.5). When all FAA were simultaneously incubated in solution, the observed capacity of each microbial isolate to metabolize them was greatly dependent on pH. At the highest pH tested (i.e., 7.5), the catalytic power reflected mainly in transamination reactions; Gln and Phe increased their concentration with time when incubated with all 12 isolates, and Met increased in concentration with eight isolates, whereas Ser, Arg, and Ile decreased. At pH 4.5, Phe accumulated, while Thr, Gly, Met, Arg, Pro, Asn, and Asp were depleted. At pH 6.0, Asn, Asp, Thr, Leu, Pro, Met, and Glu were degraded, whereas Ala built up; Ser, Ile, and Arg were completely degraded at this pH. Overall, most degradation took place at pH 6.0, which lies in the vicinity of the pH optima for most microbial enzymes (and is also close to the physiological pH). In addition, the FAA most extensively degraded are those possessing a hydrophilic character (i.e., Glu, Gln, Asn, Asp, Lys, and Arg), and which therefore dissolve better in water-based buffers. It is interesting to note that Pro, Val and Leu, all highly hydrophobic FAA, were also extensively degraded, perhaps owing to their predominance in the pool of AA residues accounted for by eve’s caseins in Serra da Estrela cheese.

The effect of α-ketoglutarate was also evaluated at pH 6.0 (Fig. 2); this compound was included as it plays an important role in the initial steps of transamination (Tamman et al., 2000). The effect of the addition of α-ketoglutarate (10 mM) was dependent on the isolate in question. As suggested by Yvon et al. (1998), the differences found in the degradation rates using different starters had to do with the concentration of substrate FAA (including glutamate). Furthermore, they concluded that FAA degradation depends also on the...
In vitro degradation of free amino acids in cocktail, brought about by four isolates (ESB 321 = Lactobacillus paracasei ssp. paracasei, ESB 327 = Leuconostoc mesenteroides ssp. dextranicum, ESB 331 = Lactococcus lactis ssp. lactis, and ESB 338 = Enterococcus faecium) from Serra da Estrela cheese, by 7 d of incubation at several pH values, viz. 4.5 (■), 6.0 (lightly shaded bar) and 7.5 (■).

Figure 1. In vitro degradation of free amino acids in cocktail, brought about by four isolates (ESB 321 = Lactobacillus paracasei ssp. paracasei, ESB 327 = Leuconostoc mesenteroides ssp. dextranicum, ESB 331 = Lactococcus lactis ssp. lactis, and ESB 338 = Enterococcus faecium) from Serra da Estrela cheese, by 7 d of incubation at several pH values, viz. 4.5 (■), 6.0 (lightly shaded bar) and 7.5 (■).

aminotransferase activities of the various starters. When isolate ESB 331 was used, it promoted degradation of almost all FAA. No observable increase in FAA uptake was detected in the presence of α-ketoglutarate when isolates ESB 323 and ESB 338 were used, except for Phe (isolate 323) and Glu (isolate 338). When isolate ESB 330 was used, the addition of α-ketoglutarate promoted uptake of Thr, Pro, Leu, Glu, and Trp. These distinct observations may be attributable to metabolic differences between the genera tested. Substrate utilization by nonstarter LAB was also studied by Tamam et al. (2000) using Biolog MT microplates; these authors confirmed that many of those microorganisms have the ability to generate energy from FAA catabolism, but only in the presence of exogenously supplied α-ketoglutarate, hence confirming the findings by Yvon et al. (1998). Similarly to what happened in individual FAA incubations, Glu and Ala were the most degraded FAA (see Figure 2), while Thr and Met tended to build up (i.e., negative values on the y-axis) in this system. The viable numbers remained relatively high during the course of the experiment at pH 7.5: initially (i.e., right after inoculation) in the log range of 9.0 to 9.5 CFU/ml, and finally (i.e., by 7 d) in the log range 8.8 to 9.2, for all strains. At pH 4.5 and 6.0, the log viable numbers per milliliter dropped slightly from 8.8 to 7.0 in the same time frame. Autolytic experiments (data not shown) encompassing the four isolates (ESB 321, ESB 327, ESB 331, and ESB 338) showed that lysis in general occurred later at more extreme pH values; this observation was quite apparent for isolate ESB 338. At pH 6.0, isolates ESB 327 and 331 lysed earlier than the other two, which might explain why isolate ESB 327 produced the largest number of volatiles via FAA catabolism (Table 4).

The enzymatic conversion of FAA is a rate-limiting factor in the development of flavor compounds during cheese maturation (Christensen and Reineccius, 1995). As demonstrated by Wallace and Fox (1997), the addition of FAA to Cheddar cheese did not result in significant flavor development. Yvon et al. (1997) suggested that transamination is the first step in the aromatic FAA degradation by lactococci, since no oxidative deamination or decarboxylation was detected. Aminotransferases, the enzymes able to catalyze these reactions, are also responsible for the uptake of Leu and Met. Gummalla and Broadbent (1999), in a study on Trp catabolism by lactobacilli, indicated that these bacteria catabolize Trp to indole-3-lactic acid through successive transamination reactions. When studying L-Phe catabolism by microorganisms originated in cheese, Lee and Richard (1984) found that phenylpyruvic acid is an intermediate in bacteria, although yeasts are able to directly produce phenylethyl alcohol from Phe; hence,
transamination (leading to formation of the corresponding keto acid) would in fact be the first step in Phe catabolism.

### Metabolism of Free Amino Acids in Curdled Milk

Table 2 depicts the amount of FAA remaining in curdled milk after 45 d of incubation with each of the four isolates tested, and corresponding standard deviation (no significant differences were found between the samples taken at 0, 15, and 30 d). Analysis of variance, at the 5% level of significance, using isolate, incubation time, and addition of $\alpha$-ketoglutarate as main factors, allowed one to conclude that those factors were always statistically significant toward degradation of all FAA, except Gln, Glu, Met, and Lys. For Gln and Glu, the addition of $\alpha$-ketoglutarate was not significant, whereas the amount of Glu, Met, and Lys degraded was not dependent on the isolate used as inoculum. Interestingly, Glu was again among the most degraded FAA in this system, whereas Ala (unlike what happens in the buffer system) was one of the least degraded FAA. The log numbers of viable cells/ml remained essentially constant (ca. 8.5) from the time of inoculation up to 45 d.

Principal component analysis of our set of data led to identification of four clear clusters (Figure 3): one comprising the 0-d samples (with and without $\alpha$-ketoglutarate) (cluster 1), a second one comprising the 30- and 45-d samples with $\alpha$-ketoglutarate (cluster 2), a third one comprising the 15-d samples with $\alpha$-ketoglutarate added (cluster 3), and the last one comprising the 15-, 30-, and 45-d samples without $\alpha$-ketoglutarate (cluster 4). Over 85% of the total variance was accounted for by the first four principal components (PC1, PC2, PC3, and PC4). The correlation matrix shows that all samples from cluster 1 are significantly (and negatively) correlated with factor 2, PC2; Pro and Lys are, on the other hand, significantly correlated (positively and negatively, respectively) with this factor. Figure 4 shows the FAA distribution along PC1 and PC2 for all samples: Asn, Asp, Thr, Ala, Met, Val, and Ile are negatively correlated with PC1 and with samples from cluster 4, meaning that samples without $\alpha$-ketoglutarate yielded high amounts of these FAA (therefore, having degraded them to a lesser extent), unlike the samples in cluster 2 (located in the diagram in an opposed position), which are those with $\alpha$-ketoglutarate and lower amounts of the aforementioned FAA; Pro and Lys correlated with PC2 and with samples from cluster 1, thus indicating that these samples had high amounts of Lys and Pro. The highest score on PC1 was obtained by Glu; this FAA, which may be produced by the action of rennet, is predominant in various types of cheese, and is known to enhance cheese flavor.
Production of Volatile Compounds from FAA

Volatile compounds present in ripened cheese are shown in Table 3. The most abundant volatile compound in Serra da Estrela cheese is 3-methylbutyric acid (Dahl et al., 2000), which is the product of metabolic breakdown of Leu, a rather abundant FAA in that cheese. However, the profile of the FAA pool in cheese derived from casein does not apparently constrain the pattern of volatiles generated: remember that the most abundant FAA in ripened Serra da Estrela cheese are Leu, Val, Glu, and Lys, but the volatile products found only in cheese (Table 3) do originate from reactions that depart from Ala, Glu, Gly, Asp, Leu, Lys, Phe, and Tyr as substrates.

The volatiles generated in vitro by catabolism of the FAA cocktails, at pH 6.0 and by 7d of incubation, by each of the four bacterial genera are listed in Table 4. Considering in detail the (known) molecular structures of such volatiles, and comparing with previous work by several authors (e.g., Engels and Visser, 1994; Wallace and Fox, 1997), one can draw a few conclusions on their putative precursors. First, it seems that these volatiles build-up via decarboxylation, deamination, transamination, and Strecker degradation reactions undergone by FAA. Subsequent processes encompass reduction (or oxidation) of Strecker degradation products, rearrangements, and cyclizations, as well as reactions between secondary products of FAA catabolism (Molimard and Spinnler, 1996). Furthermore, compounds that are typically generated in cheese were not produced by our isolates in vitro, such as ethanol, 1-propanol, butyl lactate, 4-methylvaleric acid, and phenylethyl alcohol. These pieces of evidence are obviously...
Table 4. Volatile compounds present in the FAA cocktail, inoculated with the various genera studied and incubated at 30°C for 7 d.

<table>
<thead>
<tr>
<th>Volatile compound</th>
<th>Putative precursors(s)</th>
<th>Lactobacillus</th>
<th>Leuconostoc</th>
<th>Lactococcus</th>
<th>Enterococcus</th>
</tr>
</thead>
<tbody>
<tr>
<td>Isopropylpropionate</td>
<td>Ala</td>
<td></td>
<td></td>
<td></td>
<td>x</td>
</tr>
<tr>
<td>Propionic acid</td>
<td>Ala</td>
<td>x</td>
<td>x</td>
<td>x</td>
<td>x</td>
</tr>
<tr>
<td>Ketobutyric acid</td>
<td>Glu</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Acetic acid</td>
<td>Gly</td>
<td>x</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>2-Methyl-1-butanol</td>
<td>Ile</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>3-Methyl-1-butanol</td>
<td>Leu</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>3-Methylbutyric acid</td>
<td>Leu</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Dimethyl trisulfide</td>
<td>Met</td>
<td>x</td>
<td></td>
<td></td>
<td></td>
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<tr>
<td>Benzaldehyde</td>
<td>Phe</td>
<td>x</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Acetophenone</td>
<td>Phe</td>
<td>x</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Benzoic acid</td>
<td>Phe</td>
<td></td>
<td></td>
<td></td>
<td>x</td>
</tr>
<tr>
<td>Benzoephonene</td>
<td>Phe</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Diphenylether</td>
<td>Phe/Tyr</td>
<td></td>
<td></td>
<td></td>
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<tr>
<td>2-Ethoxyethanol</td>
<td>Ser</td>
<td>x</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>2-Ethoxyethoxy-2-ethanol</td>
<td>Ser</td>
<td>x</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Glycol</td>
<td>Ser</td>
<td>x</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>2,3-Butanediol</td>
<td>Thr</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>1,2-Propanediol</td>
<td>Thr</td>
<td></td>
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<td></td>
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<tr>
<td>Indole</td>
<td>Trp</td>
<td>x</td>
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<td></td>
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<tr>
<td>Phenol</td>
<td>Tyr</td>
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<tr>
<td>2-Methyl-1-propanol</td>
<td>Val</td>
<td></td>
<td></td>
<td></td>
<td>x</td>
</tr>
<tr>
<td>2-Methylpropionic acid</td>
<td>Val</td>
<td></td>
<td></td>
<td></td>
<td>x</td>
</tr>
<tr>
<td>Isopropyl isobutyrate</td>
<td>Val</td>
<td></td>
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<td></td>
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</tbody>
</table>

X = present.

the result of our deliberate choice of a simplified, well-defined system to study a process (i.e., FAA catabolism) that is highly complex, so extrapolation to cheese should be done with care.

*Leuconostoc* strains are the most active in generation of volatiles; examples include α-ketobutyric acid, benzoic acid, and 1,2-propanediol. The volatile patterns produced by the four genera of microorganisms reflect different enzyme activities, hence suggesting that they may follow distinct metabolic pathways to generate volatiles from a given precursor.

**CONCLUSIONS**

With respect to individual FAA uptake, our experimental observations did not unfold clear relationships between those degraded preferentially by the strains tested and those most abundant in cheese. All microbial isolates exhibited a higher capacity to take up FAA at pH 6.0 than at pH 4.5 or 7.5, thus suggesting that the optimum pH for most endocellular enzymes involved in the process should be ca. 6.0. However, the pattern of volatiles formed is greatly dependent on the genus in question, owing to the distinct enzyme machineries available. Furthermore, not all volatiles formed correlate well with disappearance of their (putative) FAA precursors.

Metabolism of all FAA together in solution by the strains tested correlates poorly with the concentrations of volatiles found in cheese; realization that solubility in a matrix containing fat globules may contribute to easier access to the active centers of hydrophobic enzymes might help explain (at least partially) such evidence. Furthermore, the higher complexity of the cheese matrix relative to well-defined, homogeneous solutions may dictate breakdown and generation of specific compounds, as a result of specific additional (or alternative) pathways. Among the strains tested, *Leuconostoc* spp. were the most active in terms of conversion of FAA to volatiles.

Upcoming studies should to advantage focus on the evaluation of the enzyme machinery of selected isolates.
upon each (and all) FAA, in order to map the anabolic pathways encompassing every volatile of interest. However, our study already suggests that the wild strains tested offer great potential as tentative nonstarter cultures for enhancing the aroma in Serra da Estrela (and similar) cheeses.

ACKNOWLEDGMENTS

The authors are grateful to the European Commission for partial funding through projects FAIR CT 97-3173 entitled Control, acceleration and diversification of cheese flavor formed by enzymatic conversion of AA, and MCFI-2000-02049 entitled TRADEOFF: TRAdditional Dairy Technology toward safer Foods. Personal financial support for authors Dahl and Tavaria was provided by a Post-Doc fellowship (PRAXIS BPD-11838/97) and a Ph.D. fellowship (PRAXIS BD-9572/96), respectively, both granted by the Portuguese Government (Fundação para a Ciência e Tecnologia, PRAXIS XXI, Portugal).

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